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PLAY. Match. Gravity.

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Guide for Chapter 25.

Terms in this set (37)

Neutron Ejection. when

a neutron is emitted

from the nucleus.  ${}^1_0n$ .

Particle for Neutron

Ejection.  ${}^5_2\text{He} \rightarrow {}^1_0n +$

${}^4_2\text{He}$ .

## **Chapter 25 - Nuclear Chemistry**

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Chemistry (12th

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Textbook Authors:

Wilbraham, ISBN-10:

0132525763, ISBN-13:

978-0-13252-576-3,

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### **Nuclear Chemistry ...**

Gravity. Created by. mgray97PLUS. Chapter 25 of Prentice Hall Chemistry Vocabulary and other vocab relating to nuclear chemistry. Terms in this set (35)

radioactivity. the process by which materials give off rays emitted by uranium atoms. radiation. the penetrating rays and particles emitted by a radioactive source.

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trace the history of  
nuclear chemistry from  
dis-covery to

application. You will

identify types of  
radioactive decay and

solve decay rate

problems. You will

describe the reac-tions

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involved in nuclear fission and fusion. You will learn about applications of nuclear reactions

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Chemistry . Isotope

Examples • Prob: -An

atom of Kr has a mass

of 94 AMU. How many

protons & neutrons

does it have? •

Answer: Kr = element

#36 (36 protons).

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Neutrons = mass -  
protons, =  $94 - 36 = 58$   
neutrons. • Prob: An  
atom has 20 protons  
and 22 neutrons. Draw  
the

### **Chapter 25 - Nuclear Chemistry**

The production of  
energy in a nuclear  
reactor can be stopped  
by pulling out all  
control rods. A breeder  
reactor produces more  
fuel than it uses. The  
fission products

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produced in nuclear power plants are not radioactive. An uncontrolled chain reaction led to the nuclear accident in Chernobyl, Ukraine.

Chemistry: Matter and Change Chapter 25

149

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(show all your works and measure that you answer all questions

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from part a to part d.

thanks you) a. what is

a nuclear reaction b.

list 5 characteristics of  
nuclear reactions that  
differ from che ...

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Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic

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idea of nuclear structure, that the nucleus of an atom is composed of protons and, with the exception of  ${}^1_1\text{H}$ , neutrons.

### **25.1: Radioactivity - Chemistry**

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Chapter 25: Nuclear Chemistry Chapter  
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CHAPTER 25 What

You'll Learn You will

trace the history of  
nuclear chemistry from  
dis-covery to

application. You will

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identify types of radioactive decay and solve decay rate problems. You will describe the reactions involved in nuclear fission and fusion.

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Problems Answer  
Key**

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Chemistry 25.1 Nuclear  
Radiation 25.2 Nuclear  
Transformations 25.3  
Fission and Fusion 25.4



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Radiation in Your Life.

25. Nuclear Chemistry

25.1 Nuclear Radiation

25.2 Nuclear

Transformations 25.3

Fission and Fusion 25.4

Radiation in Your Life.

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692 Chapter 16

Nuclear Chemistry 16.1

The Nucleus and

Radioactivity Our

journey into the center

of the atom begins

with a brief review. You

learned in Chapter 3

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that the protons and neutrons in each atom are found in a tiny, central nucleus that measures about  $1/100,000$  the diameter of the atom itself. You also learned

## **Chapter 16 Nuclear Chemistry**

Chapter 25 - Nuclear Chemistry

Radioactivity

- Radioactivity is the process by which nuclei emit particles

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and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

## **Chapter 25 Nuclear Chemistry Guided Reading And Study**

...

Chapter 25 "Nuclear

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### Nuclear Chemistry

Chemistry". Use these activities to learn the vocabulary and major concepts presented in this chapter. several layers of photographic film covered with black light-proof paper encased in a plastic or metal holder. This activity was created by a Quia Web subscriber.

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Answer Key

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Name \_\_\_\_\_ Class \_\_\_\_\_

Date \_\_\_\_\_ Chapter 10

Nuclear Chemistry

Section 10.1

Radioactivity (pages  
292-297) This section  
discusses the different  
types of nuclear  
radiation and how they  
affect matter.

**Chapter 10 Nuclear  
Chemistry Section**

*Page 23/25*

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#### **10.1 Radioactivity ...**

806 Chapter 25

Nuclear Chemistry

Figure 25-2 Both Pierre and Marie Curie played important roles in

founding the field of nuclear chemistry

Marie Curie went on to show that unlike

chemical reactions, radioactivity is not

affected by changes in physical conditions

such as temperature

and pressure She is the

only person in history



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