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12:07 PM Page vi. CHAPTER 1 REVIEW Matter and Change SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. ... SHORT ANSWER Answer the following questions in the space provided. 1.

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Explain your answer. The forward reaction is favored. The weaker acid and weaker base are produced in the forward reaction.  $\text{HCO}_3^-$  competes more strongly with  $\text{H}_2\text{O}$  to donate a proton, and  $\text{OH}^-$  competes more strongly with  $\text{CO}_3^{2-}$  to acquire a proton, causing the forward reaction to be favored. MODERN CHEMISTRY ACIDS AND BASES 121

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